

Communications to the editor

STUDIES ON THE BIOSYNTHESIS OF
PENTALENOLACTONE. III.¹⁾
ISOLATION OF A BIOSYNTHETIC
INTERMEDIATE HYDROCARBON,
PENTALENENE

Sir:

Pentalenolactone is an antibiotic active against Gram-positive and negative bacteria as well as fungi²⁾ and produced by several species of *Streptomyces* such as *Streptomyces chromofuscus**, *S. griseochromogenes*** and *S. baarnensis* (an AA-57 producing organism)^{3)***}. Recently its mechanism of action was reported⁴⁾ to inhibit the enzyme glyceraldehyde-3-phosphate dehydrogenase.

During biosynthetic studies of this antibiotic, we have isolated pentalenolactone G⁵⁾, H¹⁾ and pentalenic acid¹⁾ as acidic biosynthetic intermedi-

ates of pentalenolactone (Fig. 1) from the filtered fermentation broth of *S. chromofuscus*. More recently, pentalenolactone E was obtained by CANE *et al.*⁶⁾ As a next step to shed light on the pathway from unknown precursors to pentalenic acid, a screening was carried out for less oxidized (neutral) metabolites in the mycelia of the above producing organisms. This resulted in the isolation from *S. griseochromogenes* of a sesquiterpene hydrocarbon named pentalenene (I)†.

Isolation procedures of I were as follows. *S. griseochromogenes* was cultivated for 60 hours in a jar fermentor containing a medium used for preparation of AA-57³⁾ and the fermentation broth was filtered. The mycelial cake obtained was extracted with 60% aqueous acetone and after removal of the solvent under reduced pressure, the residual solution was extracted with benzene. The benzene layer was concentrated to

Fig. 1.

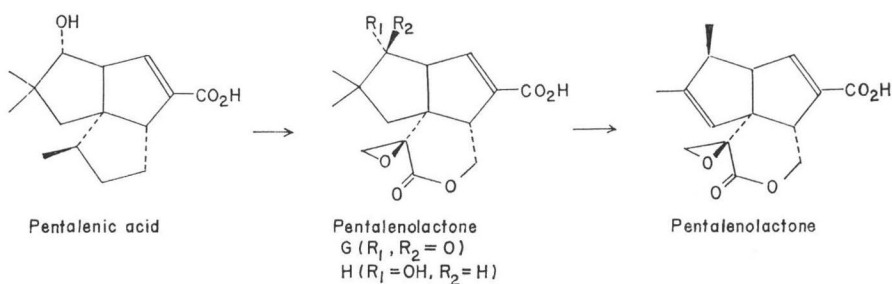
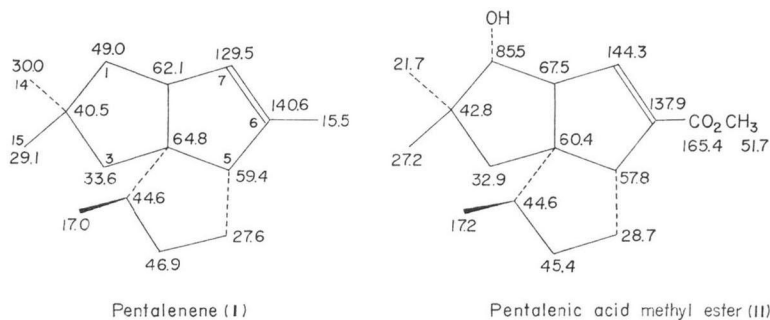


Fig. 2. ¹³C Chemical shifts of pentalenene and the methyl ester of pentalenic acid.

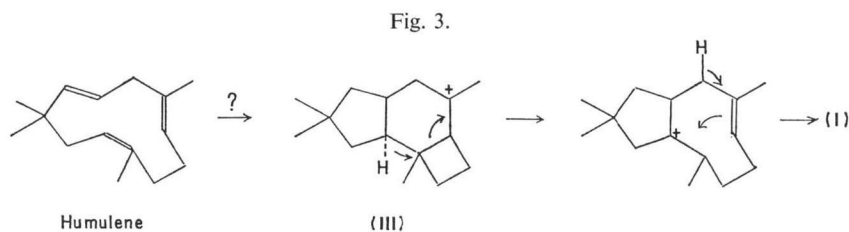


* The organism used in our previous studies^{1,5)} was classified as described herein.

** This organism, formerly called *Streptomyces* sp. 661¹⁾, was isolated by Dainippon Pharmaceutical Co. (A. TAMURA, unpublished data).

*** Production of pentalenolactone together with AA-57 by this organism has been confirmed (H. SETO and S. AIZAWA, unpublished data).

† The term pentalenene is proposed for the saturated hydrocarbon.



a small volume and *n*-hexane was added. After removal of precipitates by filtration, the filtrate was applied to a silica gel column and developed with *n*-hexane. Fractions giving a single peak by gas chromatographic analysis (retention time 7.2 minutes, column 1.5% OV-1 on Shimalite, 0.4 cm × 1 m, 100°C, flow rate of N₂ gas, 40 ml/minute) were combined and concentrated to give a pure sample of I, oil, C₁₅H₂₄ (M⁺, *m/e* found 204.1946, calcd. 204.1876), [α]_D²⁵ + 11.8° (*c* 6.8, CHCl₃).

The ¹H-nmr spectrum of I in CDCl₃ showed the presence of a *sec*-methyl (0.88 ppm, 3H, d, *J* = 7.0 Hz), two geminal methyls (0.97, 6H, s), an allylic methyl (1.60, 3H, very broad s), two allylic methines (~2.25, 2H, unresolved multiplets) and an olefinic proton (5.13, 1H, unresolved multiplet).

The ¹³C-nmr spectrum of I revealed the following carbons: 4 × CH₃, 4 × CH₂, 3 × CH, 2 × $\overset{\text{C}}{\text{---}}$ and ---HC=C--- . Comparison of the ¹³C-nmr spectra of I and pentalenic acid methyl ester¹⁾ strongly supports the structure of I as shown in Fig. 2. The downfield shift of C-14 by 7~8 ppm in I was caused by the lack of a hydroxy group present at C-1 in II. The upfield shift of C-7 in I by 15 ppm reflects the structural change of an α,β -unsaturated carbonyl system in II to an isolated double bond system in I. Further structural evidences were obtained by direct comparison of natural and synthetic samples⁷⁾ which were completely identical in physicochemical properties. MATSUMOTO *et al.*⁷⁾ had synthesized I by formolysis of protoilludyl cation equivalents such as III and proposed the following mechanism for the formation of I.

The isolation of I from a petalenolactone producing organism may be taken as strong evidence suggesting the intermediacy of humulene in the biosynthesis of pentalenolactone.

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